CHEM 131 Quiz 5 – October 2, 2019

Name ____

Complete the following problems. Write your final answers in the blanks provided. You must show your work to receive full credit. Show your answers to the correct number of significant figures with the correct units.

- 1. Answer the following questions about cyanide ion (CN^{-}) .
 - a. Complete the MO diagram below for cyanide. You may assume that the distribution of molecular orbitals is similar to that in N₂. (6 points)



b. What is the bond order for CN^{-2} (2 points)

- c. Is CN⁻ paramagnetic? Why or why not? (3 points)
- d. Would you expect the CN⁺ ion to be more or less stable than CN⁻? Why? (3 points)

2. Use molecular orbital theory to demonstrate that Be₂ is likely to be an unstable species. Use an MO diagram and bond order calculation to justify your answer. (8 points)

3. Label the molecular orbitals below as either σ_{2p} , σ_{2p}^{*} , π_{2p} or π_{2p}^{*} . (4 points)

Orbital		
Label		

IA																	VIIIA
1	1	Periodic Table of the Elements											8A 2				
н	2					i chi	oune				ients	13	14	15	16	17	He
Hydrogen	IIA											ША	IVA	VA	VIA	VIIA	Helium
1.008	2A											3A	4A	5A	6A	7A	4.003
3	4											5	6	7	8	9	10
Li	Be											B	C	N	0	F	Ne
Lithium 6.941	Beryllium 9.012											Boron 10.811	Carbon 12.011	Nitrogen 14.007	Oxygen 15.999	Fluorine 18.998	Neon 20.180
11	12	ĺ										13	14	15	16	17	18
Na	Ma	3	4	5	6	7	8	9	10	11	12	Δ	Si	P	S	CI	Δr
Sodium	Magnesium	шв	IVB	VB	VIB	VIIB	°	— vín —		IB	ШВ	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon
22.990	24.305	3B	4B	5B	6B	7B	4	8	¥	1B	2B	26.982	28.086	30.974	32.066	35.453	39.948
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca	SC	II	V	Cr	Mn	⊦e	Co	NI	Cu	Zn	Ga	Ge	AS	Se	Br	Kr
Potassium 39.098	Calcium 40.078	Scandium 44.956	Titanium 47.867	Vanadium 50.942	Chromium 51.996	Manganese 54.938	Iron 55.845	Cobalt 58.933	Nickel 58.693	Copper 63.546	Zinc 65.38	Gallium 69.723	Germanium 72.631	Arsenic 74.922	Selenium 78.971	Bromine 79.904	Krypton 83.798
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Aa	Cd	In	Sn	Sb	Te	I	Xe
Rubidium	Strontium	Yttrium	Zirconium	Niobium	Molybdenum	Technetium	Ruthenium	Rhodium	Palladium	Silver	Cadmium	Indium	Tin	Antimony	Tellurium	Iodine	Xenon
85.468	87.62	88.906	91.224	92.906	95.95	98.907	101.07	102.906	106.42	107.868	112.414	114.818	118.711	121.760	127.6	126.904	131.294
" ~ ~	Do	57-71	"LLE	¹³ To	⁴ \ A /	^{'3} Da	°~~	″ T	¹⁰ D4	^s A	° Lla	°' тı	°2	° D:	° ⁴ Da	05 A 1	°D
CS	Da		ΠΤ	la	VV	Re	US	IL	Pt	Au	пg		PD	DI	PO	At	RN
132.905	137.328		178.49	180.948	183.84	186.207	190.23	192.217	195.085	196.967	200.592	204.383	207.2	208.980	[208.982]	209.987	222.018
87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118
Fr	Ra		Rf	Db	Sq	Bh	Hs	Mt	Ds	Rq	Cn	Nh	FI	Mc	Lv	Ts	Oq
Francium	Radium		Rutherfordium	Dubnium	Seaborgium	Bohrium	Hassium	Meitnerium	Darmstadtium	Roentgenium	Copernicium	Nihonium	Flerovium	Moscovium	Livermorium	Tennessine	Oganesson
223.020	220.025		[201]	[202]	[200]	[204]	[209]	[2/0]	[201]	[200]	[203]	[200]	[203]	[203]	[293]	[254]	[294]
		57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	
	Lanth	anide	a (e I	Pr N	ld P	mS	m E	u G	id 🛛 T	ъ Г)v F	lo E	Er T	ím ۱	/b I	u
	Ser	Lant	thanum Ce	erium Prase	odymium Neod	ymium Prom	ethium San	narium Euro	opium Gado	olinium Ter	rbium Dysp	orosium Hol	mium Erl	bium Th	ulium Ytte	erbium Lut	tetium
		13	8.905 14	0.116 14	0.908 144	.243 14	1.913 1	0.36 151	1.964 15	57.25 15	8.925 16	2.500 16	4.930 16	7.259 16	8.934 17	3.055 17	4.967
	Actin	nide 89	1	-L ⁹¹	92	93	94 I	95 A	96	97 5	98	~£ 99		101		103	
	Ser	ies F	AC I	n F	al	ח ן נ	ib F	u A	mC	mt	SK C	-T I	:S F	m N			
		Act 22	7.028 Th	2.038 Prota 2.3	1.036 Urai 238	Nept 1.029 23	24 vinium 24	0nium Ame 4.064 243	3.061 Cu 24	7.070 Ber 24	Kelium Calif 7.070 25	ornium Einst 1.080 [3	254] Fer 254] 25	mium Mend 7.095 2	58.1 No 25	59.101 Law	262]

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