## CHEM 131 Quiz 5 – October 2, 2019

Name \_\_\_\_

Complete the following problems. Write your final answers in the blanks provided. You must show your work to receive full credit. Show your answers to the correct number of significant figures with the correct units.

- 1. Answer the following questions about cyanide ion  $(CN^{-})$ .
  - a. Complete the MO diagram below for cyanide. You may assume that the distribution of molecular orbitals is similar to that in N<sub>2</sub>. (6 points)



b. What is the bond order for CN<sup>-</sup>? (2 points)

Bond order =  $\frac{1}{2}$ (# electrons in bonding MO's - # electrons in antibonding MO's) =  $\frac{1}{2}(10-4) = 3$ 

c. Is CN<sup>-</sup> paramagnetic? Why or why not? (3 points)

There are no unpaired electrons, so it is not paramagnetic.

d. Would you expect the CN<sup>+</sup> ion to be more or less stable than CN<sup>-</sup>? Why? (3 points)

 $CN^+$  would have two fewer electrons, with each  $\pi_{2p}$  bonding orbital half-filled. This would have a smaller bond order and likely be less stable.

2. Use molecular orbital theory to demonstrate that Be<sub>2</sub> is likely to be an unstable species. Use an MO diagram and bond order calculation to justify your answer. (8 points)

	↑↓	
_↑↓_	σ <sup>*</sup> 2s	_↑↓_
2s		2s
	↑↓	
	σ <sub>2s</sub>	
	↑↓	
_↑↓_	σ <sup>*</sup> 1s	_↑↓_
<b>1</b> s		1s
	↑↓	
	<b>σ</b> 1s	
AO	МО	AO
	$ \begin{array}{c} \uparrow \downarrow_{-} \\ 2s \end{array} $ $ \begin{array}{c} \uparrow \downarrow_{-} \\ 1s \end{array} $	$ \begin{array}{c} \underline{\uparrow} \downarrow \underline{} \\ \sigma^{*}_{2s} \\ 2s \\ & \underline{} \\ \mathbf{a} \\ \mathbf{b} \\ 2s \\ \mathbf{b} \\ \mathbf{c} $

3. Label the molecular orbitals below as either  $\sigma_{2p}$ ,  $\sigma_{2p}^{*}$ ,  $\pi_{2p}$  or  $\pi_{2p}^{*}^{*}$ . (4 points)

Orbital				
Label	${\pmb\sigma_{2p}}^*$	$\pi_{2\mathrm{p}}$	${{f \pi_{2p}}^{*}}$	$\sigma_{2p}$

IA 1A																		VIIIA 8A
1	]					Peri	odic 1	Table	of the	e Elen	nent	S						2
H	2											13	1	4	15	16	17	He
Hydrogen	IIA											ША	IV	A	VA	VIA	VIIA	Helium
2	2A	1										3A	6	A	5A	6A	7A	4.003
1:	Bo											ĹР	ľ í	- l'	N	° ∩	Ľ E	"No
Lithium	Bendlium											D		han	Nitrogen	00000	Elucrico	INC
6.941	9.012											10.81	12.	011	14.007	15.999	18.998	20.180
11	12	]										13	14	1	5	16	17	18
Na	Ma	3	4	5	6	7	8	9	10	11	12		S	Si 📗	Ρ	S	C	Ar
Sodium	Magnesium 24 205	ШВ	IVB	VB	VIB	VIIB		— vш —		IB	ПВ	Alumin	im Sili	con	Phosphorus	Sulfur	Chlorine 25.452	Argon
19	24.505	21	48	23	24	7B 25	26	27	28	18	30	31	32	3	30.574	34	35	36
ΪK	Ca	Sc	Т	ν Γ	Cr	Mn	- Eo	6	Ni		7	Ga Ca	G	آ ما	Δc	So	Br	"Kr
Potassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron	Cobalt	Nickel	Copper	Zinc	Galliu	n Germ	anium	Arsenic	Selenium	Bromine	Krypton
39.098	40.078	44.956	47.867	50.942	51.996	54.938	55.845	58.933	58.693	63.546	65.38	69.72	3 72.	631	74.922	78.971	79.904	83.798
37	38	39	40	41	42	43	44	45	46	47	48	49	50	5	1	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cc	l   In	S	n 📗	Sb	Те	I	Xe
Rubidium 85.468	Strontium 87.62	Yttrium 88.906	Zirconium 91.224	Niobium 92.906	Molybdenum 95.95	Technetium 98.907	Ruthenium 101.07	Rhodium 102.906	Palladium 106.42	Silver 107.868	Cadmiu 112.41	m Indiur 4 114.81	n Ti 8 118	in .711	Antimony 121.760	Tellurium 127.6	Iodine 126.904	Xenon 131.294
55	56	57-71	72	73	74	75	76	77	78	79	80	81	82	8	3	84	85	86
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Διι	H	1 TI	P	h	Bi	Po	Δt	Rn
Cesium	Barium		Hafnium	Tantalum	Tungsten	Rhenium	Osmium	Iridium	Platinum	Gold	Mercu	y Thalliu	n Le	ad	Bismuth	Polonium	Astatine	Radon
132.905	137.328		178.49	180.948	183.84	186.207	190.23	192.217	195.085	196.967	200.59	2 204.38	3 20	7.2	208.980	[208.982]	209.987	222.018
87	88	89-103	104 Df	105	106	107 DL	108	109	110	"" "	112	113	114	• P	15	116	117 <b>T</b> -	118
<b>Fr</b>	Ra		RT	DD	<b>5</b> g	DN	ПS	Ινιτ	DS	ĸg					IVIC	LV	IS	Ug
223.020	226.025		[261]	[262]	[266]	[264]	[269]	[278]	[281]	[280]	n Copernic [285]	ium Ninonii [286]	m Fiero	89]	[289]	[293]	[294]	[294]
		57	58	59	60	61	62	63	64	65	6	66	7	68	69	70	71	
	Lanth	anide ies	.a   C	Ce	Pr   N	ld   P	m   S	m   E	Eu   G	id 📋	ТЬ 📗	Dy	Ho	Er	r   Ti	m    ነ	/b   l	.u
		Lant 13	hanum Ce 8 905 14	rium Prase	odymium Neod	ymium Prom	ethium San 1913 19	harium Eur	opium Gad	olinium 1	erbium	Dysprosium	Holmium 164 930	Erbiun 167.25	m Thu	ilium Ytte	erbium Lu 3.055 1	tetium 74 967
		89	90	91	92	93	94	95	96	97	9	8 9	9	100	101	102	103	1.507
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	Ser	ies Act	inium The	prium Prot	actinium Ura	nium Nept	tunium Plut	onium Am	ericium Ci	urium B	erkelium	Californium	Einsteinium	Fermiu	im Mende	elevium No	belium Law	rencium
		22	7.028 23	2.038 2	1.036 23	3.029 23	7.048 24	4.064 24	3.061 24	7.070	47.070	251.080	[254]	257.09	95 25	8.1 25	9.101	262]
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