CHEM 131 Quiz 2 – September 4, 2019

Name _____

Complete the following problems. Write your final answers in the blanks provided. You must show your work to receive full credit. Show your answers to the correct number of significant figures with the correct units.

1. Complete the table below. For each orbital indicate the number of radial and angular nodes. (10 pts)

Orbital	3р	3d
Sketch		
n		
l		
Number of Radial Nodes		
Number of Angular Nodes		

2. Draw below and orbital energy level diagram for phosphorous. Fill the orbitals with electrons following the aufbau principle and Hund's rule. How many unpaired electrons does phosphorous have? (7 points)

- 3. Write the spectroscopic notation for the electron configuration for each of the following species. (8 points)
 - a. silicon
 - b. cerium
 - c. sulfide ion
 - d. iron (III) ion

Bonus: Complete the table below: (2 pts each)

Compound Name	Formula					
copper (I) sulfate						
	$Fe_3(PO_4)_2$					

IA 1A 1 Hydrogen 1.008 3 Li Lithium 6.941 11	2 IIA 2A 4 Beryllium 9.012					Peri	odic	Table	of the	Elem	ients	13 IIIA 3A 5 B Boron 10.811 13	14 IVA 4A 6 C Carbon 12.011 14	15 VA 5A 7 Nitrogen 14.007	16 VIA 6A 8 0 Oxygen 15.999	17 VIIA 7A 9 Fluorine 18.998	VIIIA 8A 2 Heium 4.003 10 Nen 20.180 18
Na	Mg	3 ШВ	4 IVB	5 VB	6 VIB	7 VIIB	8	9 	10	11 IB	12 IIB	Aluminum	Si	Phosphorus	Sulfur	Chlorine	Ar
22.990	24.305 20	3B 21	4B	5B 23	6B	7B 25	√ 26	8	28	1B 29	2B 30	26.982	28.086	30.974 33	32.066	35.453 35	39.948 36
K Potassium 39.098	Calcium 40.078	Scandium 44.956	Titanium 47.867	Vanadium 50.942	Cr Chromium 51.996	Manganese 54.938	Fe Iron 55.845	Cobalt 58.933	Nickel 58.693	Cu Copper 63.546	Zn Zinc 65.38	Gallium 69.723	Germanium 72.631	As Arsenic 74.922	Se Selenium 78.971	Bromine 79.904	Krypton 83.798
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rubidium 85.468	Sr Strontium 87.62	Yttrium 88.906	Zr Zirconium 91,224	Niobium 92,906	Mo Molybdenum 95.95	Tc Technetium 98.907	Ruthenium 101.07	Rhodium 102,906	Palladium 106.42	Ag Silver	Cadmium 112,414	Indium 114.818	Sn 118.711	Sb Antimony 121,760	Te Tellurium 127.6	Iodine	Xe
85.468 55	87.62 56	57-71	91.224	92.906 73	74	98.907 75	76	77	106.42 78	107.868 79	80	114.818 81	82	83	84	126.904 85	131.294 86
Cesium 132.905	Ba Barium 137.328		Hafnium 178.49	Tantalum 180.948	Tungsten 183.84	Re Rhenium 186.207	Osmium 190.23	Iridium 192.217	Platinum 195.085	Au Gold 196.967	Hg Mercury 200.592	Thallium 204.383	Pb Lead 207.2	Bi Bismuth 208.980	Polonium [208.982]	At Astatine 209.987	Rn Radon 222.018
87	88	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118
Francium 223.020	Radium 226.025		Rutherfordium [261]	Db Dubnium [262]	Seaborgium [266]	Bh Bohrium [264]	Hs Hassium [269]	Mt Meitnerium [278]	Darmstadtium [281]	Roentgenium [280]	Copernicium [285]	Nh Nihonium [286]	Flerovium [289]	Mc Moscovium [289]	Lv Livermorium [293]	Ts Tennessine [294]	Oganesson [294]
	Lanth Ser	ies Lant	thanum Ce	rium Prase	Pr odymium 10.908	mium Prom	ethium Sa	marium Eur	opium Gado	linium Ter	bium Dysp	rosium Hol	lmium Erl	bium Th	ulium Ytte	rbium Lut	LU tetium 14.967
	Actir Ser	ies Act	tinium The	prium Prot	Pa L actinium Urar 11.036 238	ium Nept	tunium Plu	tonium Am	ricium Cu	rium Berk	celium Califo	ornium Einst	teinium Fer	mium Men	delevium Nol	belium Law	rencium 262]