## Summer 2011

Complete each of the following problems. For numerical problems, you must show your work in order to possibly earn full credit.

1. Identify the following as either *chemical* or *physical* changes or properties. (2 pts. ea.)

 a.	Vinegar and baking soda produce a gas when mixed.
 b.	Water boils at 100°C.
 C.	Chocolate melts if you leave it in the sun.
 d.	Cars rust after driving on salted roads.

Convert the following metric quantities. (4 pts. ea.)
a. 242 mm = ? km

b.  $0.00468 \text{ mg} = ? \mu \text{g}$ 

 You conduct an experiment where 31 grams of phosphorous is reacted with oxygen until all of the phosphorous has been converted to a phosphorous oxide compound. In the experiment, 71 g of the compound is produced. From this information, what mass of oxygen is required to produce 13 grams of this product? (8 pts.)

**Bonus:** List one example of each of the following. There are many acceptable answers to each part. (2 points each)

a. a homogeneous mixture: \_\_\_\_\_

b. a heterogeneous mixture: